# EFFECT OF INCREASING FREE SPACE ABOVE THE CHARGE ON THE COMPOSITION AND YIELD OF THE VOLATILE PRODUCTS FROM COAL CARBONIZATION

C. Ortuglio, J. G. Walters, and D. E. Wolfson

U. S. Department of the Interior, Bureau of Mines Pittsburgh Coal Research Center, Pittsburgh, Pa. 15213

#### **ABSTRACT**

The effect of increasing the free space above the charge on the composition and yields of the volatile products from the carbonization of Pittsburgh-bed coal and one industrially used blend from the eastern United States and one from western United States was investigated with the BM-AGA carbonization apparatus.

Increasing the free space above the charge from 1 inch to 3 inches, to 6 inches and to 9 inches resulted in lower tar yields and increased gas and light oil yields. Anthracene and naphthalene yields were increased, quinoline and benzene insoluble fractions of both the tar and pitch increased considerably with increased free space, although the carbon content of the tars and pitches increased only slightly.

Increased cracking, a result of the increase in free space, caused a decrease in the yields of tar acids, tar bases, neutral oils, olefins, aromatics, and paraffins and naphthenes. Gas composition was relatively constant; hydrogen and methane tended to increase and ethane decrease.

Results of the investigation are being used to design a unit to upgrade the volatile products of carbonization by cracking the hot products as they leave the carbonization chamber.

#### INTRODUCTION

The Bureau of Mines, in cooperation with the American Gas Association, jointly developed a pilot-scale carbonization test apparatus (BM-AGA) to determine the carbonization characteristics of coals and to evaluate the byproduct yields (5).

Although the BM-AGA test apparatus is used primarily to determine the coking characteristics of coals for metallurgical use, the objective of this investigation was to determine the effect of cracking on the quality of carbonization products by varying the free space above the coal charge when the temperature is kept constant.

Carbonization yields are influenced by the rank of coal carbonized, whereas their composition is primarily a function of the temperature and contact time of the evolved vapors with the heated surfaces within the retort or ovens. Decomposition of vapors proceeds in two steps: Primary decomposition of gaseous vapors as they are evolved from the plastic layer at moderate temperatures and then pass through the hot coke surfaces, and secondary decomposition as the vapors pass through the free space above the coal charge with less contact surface. Porter (4) in a study of coal carbonizing equipment has found that the variance of contact time of the gases and vapors passing through the heated spaces in approved industrial ovens may be 100 percent or more.

# EXPERIMENTAL PROCEDURES

Procedure and apparatus for BM-AGA carbonization tests have been described in detail in a previous publication (5). Basically, the apparatus consists of a cylindrical steel retort, electrically controlled resistance-type furnace, product

recovery train, gas scrubbers, meters, and sampler. The standard retort used throughout this investigation is 26 inches high and 18 inches in diameter. The results from previous carbonization tests, using standard retorts and 900°C carbonization temperatures have been correlated with industrial (7) and experimental oven (6, 8) data. The coal charges were adjusted in height so as to leave 1-, 3-, 6-, and 9-inches of free space above the coal charge. Duplicate carbonization tests were made at 900°C and the results reported are averages of two determinations.

A single coal (1166) and two commercial coal blends (1161 and 1167A) were used in this investigation. Coal 1161 was carbonized as received so that the size consist would be comparable to that charged at the coke plant. Coals for tests 1166 and 1167A were received separately in lump form and were crushed in the hammer-mill and blended.

In computing yields from BM-AGA carbonization tests, the quantitative yields are based on U. S. gallons (231 cubic inches) and short tons (2,000 pounds). Coke yields are reported as dry coke, weight-percent of coal carbonized. The yield, specific gravity (determined), and gross heating value of the gas are reported as stripped of light oil and saturated with water vapor at 60°F and under a pressure equivalent to 30 inches of mercury. Light oil refers to the crude product stripped from the gas. Liquor includes the fixed ammonia and absorbed free ammonia. Ammonium sulfate is reported in pounds per ton of coal and includes the total free and fixed ammonia. The tar yield, properties, and constituents are reported on a dry basis and includes only that light oil which condenses with the tar, and the specific gravity of the tar is reported at 15.5°C/15.5°C.

The yields of carbonization products are given on the as-carbonized basis (table 1). The latter basis is used to compare coals when the moisture and ash contents differ significantly. Comparisons lose their significance because a high percentage of ash or moisture makes the yield of coke or liquor artifically high and those of other products correspondingly low. The calculation of moisture- and ash-free basis assumes that all ash remains in the coke and that all moisture in the coal is recovered as liquor.

# Properties of Tar

The tar yields are affected by coal rank, temperature of carbonization and free space above the coal charge. The temperature of carbonization is the most significant contributing factor in determining tar quality. Increased free space permits the vapors to remain longer at the higher temperature resulting in considerable variations in tar quality.

The effect of free space on tar quality is presented in table 2. Cracking of tar due to increased free space resulted in progressive increases in its specific gravity, naphthalene yield, anthracene yield, except for the 9-inch free space and a reduction in tar acids, bases, and neutral oils. It is interesting to note that unlike naphthalene, the anthracene yield was slightly less at the 9-inch free space for all tests. There was no significant change in the residue yield.

The chemical composition of tars and pitches is presented in table 3. As expected, the carbon content increased and the hydrogen content decreased with progressive increases in the free space.

Quinoline (1) and benzene (free carbon) (5) insoluble values for the tars and pitches, as well as the softening point of the pitches were determined. These are critical factors, commercially important, in determining the suitability of coal tar pitches as electrode binders. The tar values were determined on the whole tar and the pitch values were determined on the +350°C fraction of the tar. The cube-in-air method was used to determine the softening point of the pitches. The results are presented in table 4.

Table 1. - Yields of carbonization products, as-carbonized basis

				felds,	weight	Yields, weight-percent			Y	ields per	Yields per ton of coal	
	Free										ı	Ammonium
Coal	SDACe				Light		Free		Gas	Tar		sulfate,
number	inches	Coke	Gas	Tar	011	Liquor	ammonia	Total	cu. ft.	gallons	- 1	pounds
						c	,	9	0.			
1161	_	•	18.2	ъ. С	1.13	D.	0.150	100.0	004.01	£**T	79.7	617
	က	•	16.5	5,5	1,17	10.2	.177	99.1	10,640	11,2	3,25	23.6
	9	•	16.2	5.5	1,23	10,3	.180	99,1	10,770	11.1	3.42	22.8
	்	65,5	16,3	5,3	1,13	10.0	•209	<b>h</b> *86	10,790	10.2	3,16	23,1
3311	-	7	9 11		1,28	ď	166	99,0	10.400	15.8	3.57	25.3
2011	٠,	9 4	15.3		1.24	0 0	168	99,1	10,750	16.5	3,45	23,7
	יט כ	9 4	15.3		1.31	7-6	173	98.5	10,700	15.2	3,64	22,6
	ათ	4.99	15.2	7.5	1,37	7.7	,174	98,3	10,850	14.8	3,81	22,1
	:	; ·						,		,	,	
1167A	ᆏ	70.8	13,6		1,16	7.0	. 163	<b>†</b> •66	10,140	13.7	3,23	23.2
	e	70.7	13.9		1,15	6.2	.161	98.5	10,290	12.9	3,20	20.8
٠	· vo	70.7	13.9		1.24	6.1	•16 <sup>4</sup>	98*3	10,350	12.1	3.47	20.0
	ı <b>o</b> n	70.7	14,1	6.1	1,21	6.3	.149	986	10,490	11.9	3,36	18.8

Table 2. - Specific gravity and component tar yields

					Yields	Yields, gallons per ton	per ton			lields, pounds per ton	as per con
						6	1	Neutral Oils	S		
	77.77	6:01							Paraffins		
	VOID	obeciic			Nontre				and		
Coal	Space	gravity	Acids	Bases	oils	Residue		Olefins Aromatics	Naphthenes	Naphthalene	Anthracene
i amper	11101103										
				,		0	000	2 95	0.16	5,36	2,81
1161	٦	1,17	0.52	0.33	3.50	70.0	0	7 (		C-1-	3 58
1	c	01 1	35	.27	2,89	09-9	• 31	2,52	90.	7.01	•
	2	6101	•			,	35	2.04	.02	8,70	5.25
	9	1.20	• 20	77.	7.97	01.	77.				1, 7,
		1 23	11	.15	1.63	7,03	•17	1.45	<b>*</b> 01	9,00	1,.
	n	7 * *	•								
					(			2 52	01	3.57	1,36
1166	_	1.16	. 78	.32	3.95	07.0T	66.	70.0	•		-
2011	• •		ŭ	ac	2 69	10.87	.32	3,33	₹0.	07./	7.34
		1.18	60.	07.			. 00	30.0	. 02	10.32	3.27
	ع	1.20	.25	• 24	7.5.7	10.89	67.	2.5			, 5,
			000	22	2.20	10.86	.24	1,95	•01	11,39	76.7
	'n	77.7	•	:			,				
						0	16	אור ני	60	t T t	1,92
11674	_	1.17	69	/7.	3.50	0000	10.	•			3 13
11010	4 (		22	36	7 11	8.82	. 20	2,19	£0°	60./	31.6
	יי	6T •T	· ·	07.	• • •			٦, ١	.01	8.97	3,31
	Œ	1.22	•21	• 18	86.1	g. 0.4	77.	7			20 05
	•	1 23	17	117	1.78	8.57	.21	1.57	•01	36.35	6.3

Table 3. - Effect of free space on the chemical composition of tar and pitch

		Та	r			Pi	tch		
Void space, inches	1	3	6	9	1	3	6	9	
			Coal	1161					
Hydrogen	5.7	5.2	5.0	4.7	4.8	4.4	4.3	3.9	
Carbon	90.5	91.4	92.7	93.1	92.3	92.8	93,8	94.2	
Nitrogen	1.1	1.2	1.1	1.0	1.4	1.3	1.2	1.0	
Oxygen	2.2	1.7	0.7	0.7	1.1	1,1	0,3	0.4	
Sulfur	0.4	0.4	.4	.4	0.3	0.3	• 3	.3	
Ash	.1	.1	•1	.1	•1	.1_	.1	, 2	
			Coal	1166					
Hydrogen	5.4	5,0	4.8	4.7	4.9	4.7	4.6	4.2	
Carbon	89.7	90.4	90.8	91.1	90.5	91.0	91.9	92.1	
Nitrogen	1.3	1.3	1.2	1.1	1.5	1.4	1.3	1.3	
Oxygen	2.5	2.2	2.1	2.0	2.1	2.0	1.3	1,6	
Sulfur	1.0	1.0	1.0	1.0	0.8	0.7	0.7	0.7	
Ash	0.1	0.1	0.1	0.1	• 2_	• 2	.2	.1	
	Coal 1167A								
Hydrogen	5.8	5.4	5.1	5.0	4.9	4.7	4.6	4.4	
Carbon	90.3	92.0	91.5	92.0	91.1	91.6	92.0	92.2	
Nitrogen	1.3	1.3	1.3	1.3	1.5	1.4	1.4	1.3	
Oxygen	1.4	0.4	1.0	0.6	1.7	1.5	1.2	1.3	
Sulfur	1.0	. 8	. 1.0	1.0	0.7	0.7	0.7	0.7	
Ash	0.2	.1	0.1	0.1	.1	.1	.1	.1	

Table 4. - Effect of free space above coal charge on quinoline and benzene insolubles

	Void	Ta	arl	P	itch <sup>1</sup>	Softening
Coal number	space, inches	Quinoline insolubles	Benzene insolubles	Quinoline insolubles	Benzene insolubles	point, °C
1161	1	1.70	6.61	3.85	25.95	119.0
1101	3	2.73	8.17	3.96	24.25	105.0
	6	4.84	11.29	8.87	28.33	110.0
	9	11.07	17.42	22.03	32.55	138.0
1166	1	2.26	8.64	2,82	22,55	96.8
1100	3	4.13	11.60	4.78	26.55	101.0
	6	7.29	14.26	9.14	28,33	94.0
	9	11.15	18.76	14,34	30.87	85,8
1167A	1	2.57	7.66	2,86	21.75	94.0
110/1	3	3.87	10.34	4.73	23.70	94.3
	6	6.39	12.47	7.60	25.15	82.3
	9	9.50	15.46	12.15	27.85	85.0

 $<sup>^{1}</sup>$ Weight-percent for tar and pitch.

The insolubility values for all tars and pitches increased with increasing free space. There were marked increases at the 9-inch free space for all tests, indicating the formation of insoluble high boiling hydrocarbons.

## Properties of Gas

Properties of byproduct gas vary with carbonizing conditions as well as with coal characteristics. An investigation (9) of the gas yielded by coals carbonized by the BM-AGA method at 900°C showed that the physical and chemical properties of the gas depend on the rank of coal.

The yield of carbonization gas increases with increased carbonization temperature; similarly, maintaining the effluent vapors at elevated temperatures by increasing the free space above the coal charge exposes the vapors to high temperatures for a longer time resulting in degradation of the volatile material with a still greater increase in the gas yield. Davis and Auvil (3) have attributed this increase to greater cracking of hydrocarbons in the enlarged free space. Table 1, confirms these results, in that the gas yield increases with increased free space.

The chemical and physical properties of the gas are presented in table 5. Two important properties of gas are its heating value and hydrogen sulfide content. The heating value, generally considered to be the most important property of byproduct gas, is presented on both the cubic foot and pound-of-coal basis. The greatest heat recovery for the longest exposure time of gas in the retort was 112 Btu per pound of coal (1167A) or a gain of 224,000 Btu per ton of coal. Coal 1161 showed a steady decline in heating value with increased exposure time.

The hydrogen sulfide content of the gas, calculated to grains per 100 cubic feet of gas, is becoming increasingly important because of air pollution restrictions, and must be reduced to limited concentrations before disposal. Results indicate that the formation of hydrogen sulfide is related to exposure time; however, the concentration of hydrogen sulfide in the byproduct gas is dependent on the sulfur content of the coal and is not related to coal rank. Although the sulfur content for coal 1167A (1.4 percent) was higher than that for coal 1161 (0.6 percent), the net increase in hydrogen sulfide, attributed to free-space cracking, was only 14.8 percent, compared with 26.0 percent for coal 1161.

## Light Oil

The composition of light oil, like the tar, is a function of the carbonizing conditions of which temperature, free space above the coal charge, and the contact time of the gas in the hot free space are the most important; however, the total yield of light oil under normal carbonizing conditions is dependent largely upon the rank of the coal.

Benzene, the principal constituent of the light oil, is a decomposition product whose formation is accelerated under conditions favorable to cracking effluent vapors.

The percent composition of benzene in the light oil, as presented in table 6, progressively increased with an increase in free space, whereas all other constituents of light oil were adversely affected by the free space increases.

The total light oil yield reached a maximum at the  $\delta$ -inch free space for the coal blends and at the 9-inch free space for the straight coal.

Table 5. - Chemical and physical properties of gas

	Void	Соп	position,	dry, volu	me-perce	nt	Heating	value	Hydrogen sulfide
Coal number	space, inches	Hydro- gen	Carbon monoxide	Methane	Ethane	Ethylene	Btu per cu.ft.	Btu per lb.coal	grains per 100 cu.ft.
1161	1	60.87	5.98	25.78	2.43	5.03	618	3,235	131
	3	62.13	6.07	24.51	2.15	5.14	608	3,231	157
•	6	64.01	5.07	23.79	1.60	5.53	600	3,180	157
	9	63.29	5,91	27.40	1.08	2.32	576	3,121	165
1166	1	59.38	6.49	30.08	1.42	2.63	603	3,133	638
	3	59.37	6.13	30.11	1.56	2.83	608	3,268	714
	6.	59.09	6.45	30.97	0.92	2.57	601	3,221	707
,	9	61.87	4.52	29.85	.75	2.96	596	3,189	828
1167A	· 1	63,91	4.31	28.81	1.30	1.67	583	2.956	561
	3	61.94	5.12	30.05	0.85	2.04	587	3,020	600
	6	61.98	5.15	30.08	.70	1.29	581	3,007	606
	9	61.91	4.85	30.94	.56	1.74	- 585	3,068	644

Table 6. - Composition of light oil, gallons per ton

Coal number	Void space, inches	Benzene	Toluene	m <sub>e</sub> p-xylene	o-xylene	Ethyl- benzene
1161	1	2.24	0.38	0.12	0.05	0.01
	3	2.65	.42	.14	.04	.006
	6	2.91	.40	.09	.01	.004
	9	2.89	.22	• 04	.01	-
1166	1	2.77	.64	.13	.02	•004
	3	2.79	•55	.08	• 02	.003
	6	3,13	<b>.</b> 45	.05	•01	.001
	9	3.34	.45 .05 .01 .42 .04 .01	.001		
1167A	1	3.05	•06	.10	.02	.013
	3	3.05	•05	•07	.02	• 003
	6	3.39	<b>.</b> 04	.03	•01	.001
	9	3.30	_03	.03	.01	-001

### Properties of Coke

The physical properties of the cokes were determined by standard methods of the American Society for Testing Materials (2). The tumbler indices, 1-inch stability and 1/4-inch hardness, are the only coke properties reported as these parameters are most commonly used in evaluating metallurgical coke.

Variations of free space had no effect on the coke yield. The only significant change in the coking properties was for coal 1166 in which the 1-inch stability index was progressively lowered from 33 for the 1-inch free space to 28 for the 9-inch free space. The stability and hardness indices for both coal blends remained essentially the same.

#### CONCLUSIONS

The yields of carbonization products are influenced by the rank of coal used and the temperature of carbonization. The primary decomposition is influenced by the rate of heating and the secondary decomposition by the time of contact of the vapors while in the hot retort and is dependent on the volume of free space per unit volume of charge.

The cracking of tar due to increased free space resulted in progressive increases in specific gravity, naphthalene yield, anthracene yield (except for the 9-inch free space test) and a reduction in tar acids, bases, and neutral oils.

Free space increases had no effect on the coke yield or coke properties except for the 1-inch tumbler stability for coal (1166) which was progressively lowered with increased free space.

The light oil yield was increased and reached a maximum at the 6-inch free space for the coal blends, and the 9-inch free space for the straight coal.

There was a progressive increase in gas yield, with the greatest variation of 500 cubic feet per ton of coal and the heat recovery for the longest exposure of gas in the retort was 112 Btu per pound of coal.

Increased free space above the coal charge during carbonization, permitted longer residence time of the vapors at a specific temperature and improved the quality of the byproducts. However, since coke is the major and most valuable product of coal carbonization, it would be economically infeasible to decrease the capacity of commercial ovens. In recognition of this, the Bureau of Mines is investigating the upgrading of volatile materials, in a separate cracking unit, outside of the coking furnace. Results of this investigation will be published in a subsequent report.

#### REFERENCES

- American Society for Testing Materials. Bituminous Materials, Soils, Skid Resistance, Part II, March 1969, pp. 701-704.
- 2. American Society for Testing Materials, ASTM Standards on Coal and Coke, 1954, 155 pp.
- Davis, J. D. and S. Auvil. High Temperature Carbonization of Coal. Ind. & Eng. Chem., vol. 27, April 1935, p. 459.
- 4. Porter, H. C. Improvement of Design of Coal Carbonization Equipment. Ind. & Eng. Chem., vol. 24, 1932, pp. 1363-1368.
- Revnolds, D. A., and C. R. Holmes. Procedure and Apparatus for Determining Carbonizing Properties of American Coals by the Bureau of Mines-American Gas Association Method. BuMines Tech. Paper 685, 1946, 35 pp.
- 6. Staff, Division of Bituminous Coal. Comparative Evaluation of Coking Properties of Four Coals. BuMines Inf. Cir. 8010, 1961, 22 pp.
- 7. Walters, J. G., and G. W. Birge. Duplicability Studies of the BM-AGA Test Method and Correlation with Industrial Coking. BuMines Rept. of Inv. 5467, 1959, 16 pp.
- 8. Wolfson, D. E., G. W. Birge, and J. G. Walters. Comparison of Properties of Coke Produced by BM-AGA and Industrial Methods. BuMines Rept. of Inv. 6354, 1964, 19 pp.
- 9. Wolfson, D. E., and D. A. Reynolds. Yields and Properties of Gases From BM-AGA Carbonization Tests at 900°C. BuMines Tech. Paper 693, 1946, 12 pp.